

MODELLING PESTICIDE VOLATILIZATION FROM CROP AT THE FIELD SCALE

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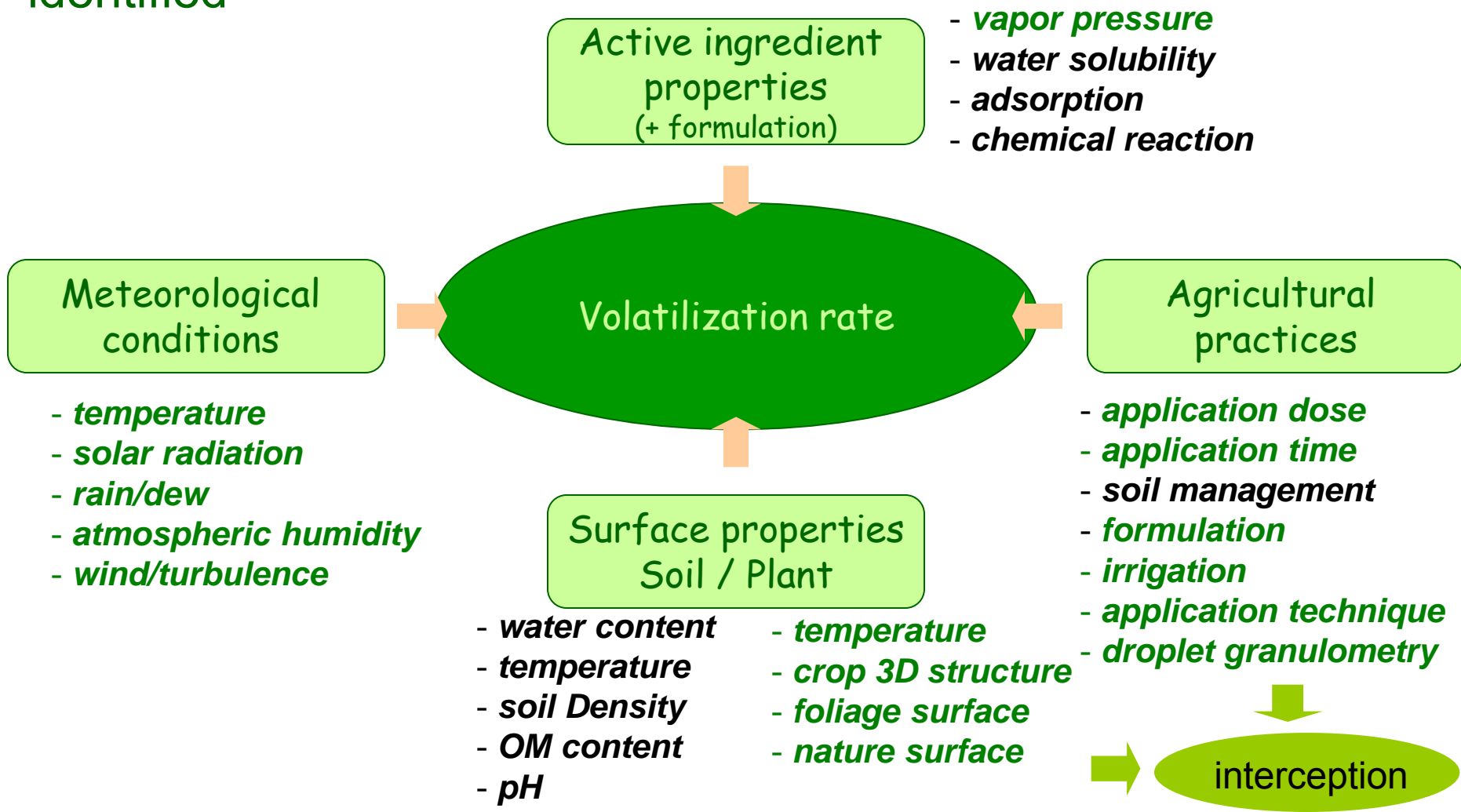
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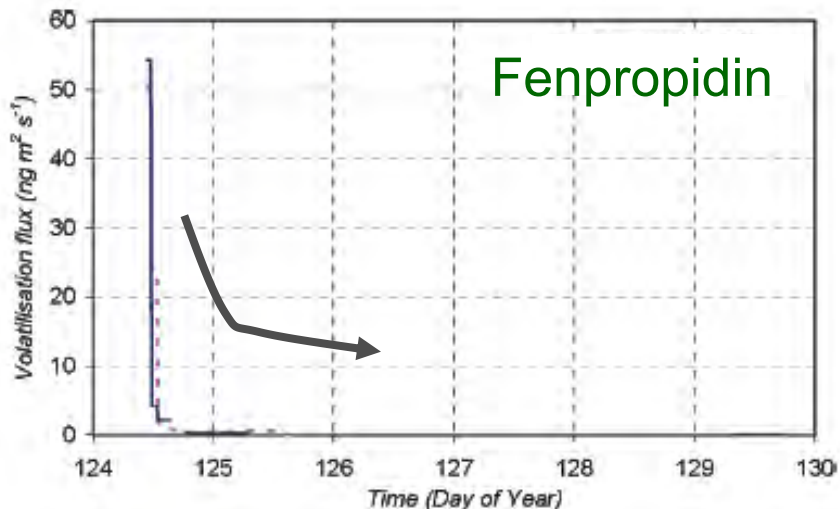
Outline

- Introduction – Context – Objectives of the study
- Materials and methods
 - ❖ Model
 - ❖ Experimental setup
- Results in terms of leaf temperature and pesticide volatilization fluxes (simulated vs measured)
- Conclusions and Perspectives

Context: Volatilization rates from crop, main involved factors identified

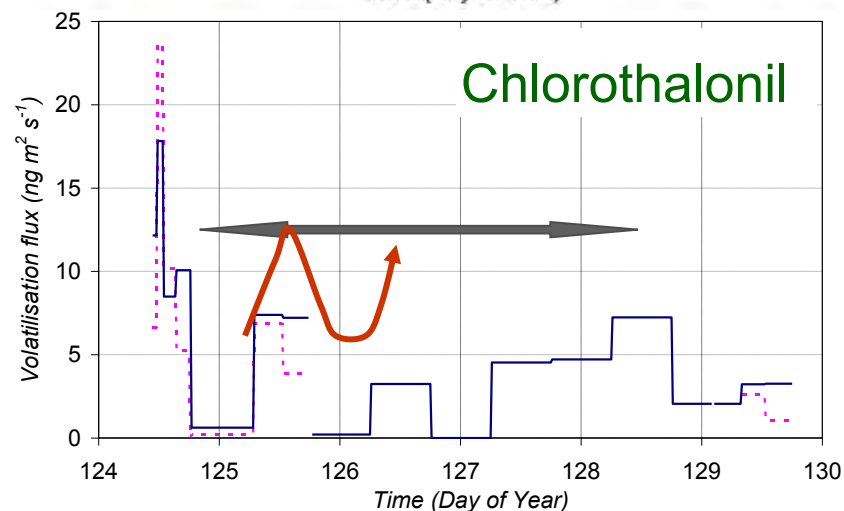


Context: Volatilization rates from crop observed at the field scale (ng/m²/s)



e.g. two fungicides of wheat

- ❖ Various orders of magnitude
- ❖ Different time dynamics
- ❖ Diurnal cycle

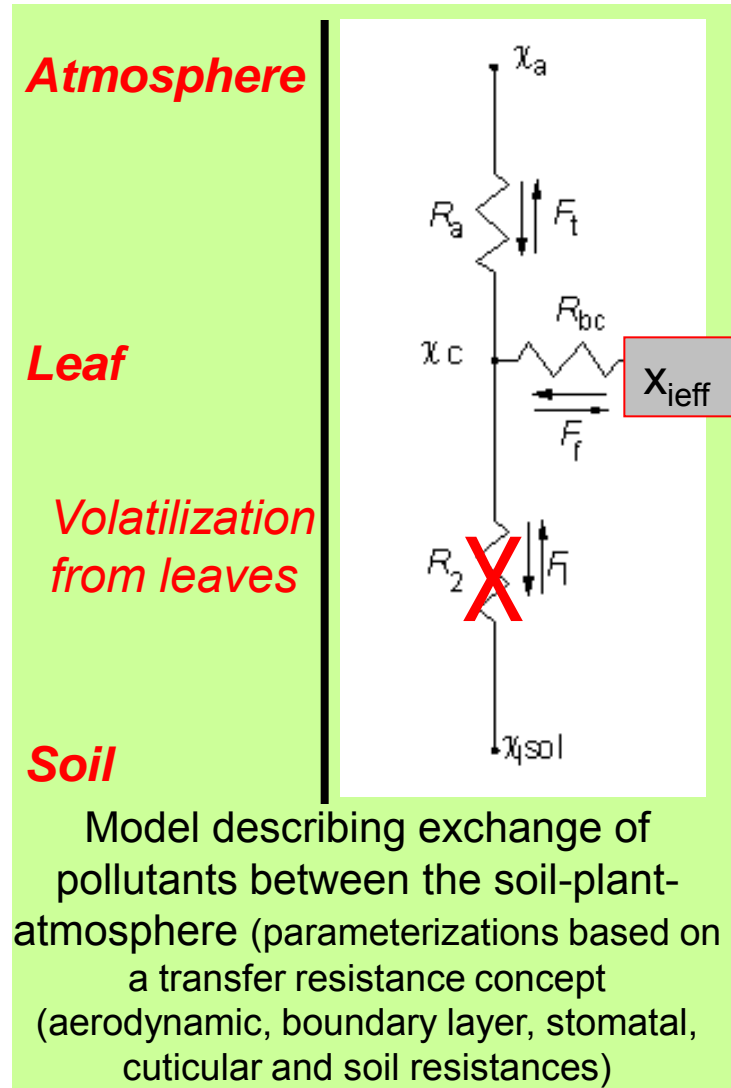


Bedos et al. (2010)

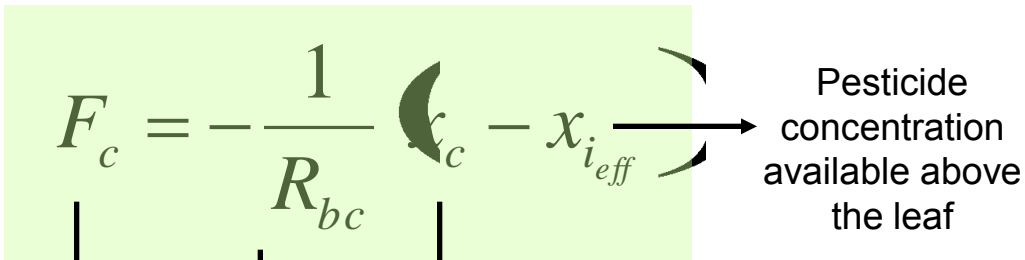
Objectives of this study

- Model the pesticide volatilization from leaf at the field scale, in a mechanistic way *i.e.* taking into account main factors involved at an adapted time scale (infra-hourly)
 - Test this model with dataset
 - Study the contribution to the global volatilization from crop of the volatilization from soil and the volatilization from leaves
- => towards an emission module to be used for modelling the pesticide behaviour in the atmosphere at larger scales

Material and Methods: The SURFATM model (Personne et al., 2009), volatilization from crop



Adaptation to pesticides (following parameterization of Leistra, et al., 2004, PEARL)



volatilization flux from leaves

Crop boundary layer resistance

Pesticide concentration within the crop air

$$x_{ieff} = x_i \frac{Q(t)}{\lambda}$$

$$x_i = \frac{P_{sat}}{RT} M$$

Competing processes

$$Q(t) = Q(t^{-1}) - (F_c - (K_{pen} + K_{deg})Q(t^{-1}))\Delta t$$

Material and Methods: The SURFATM model, other processes

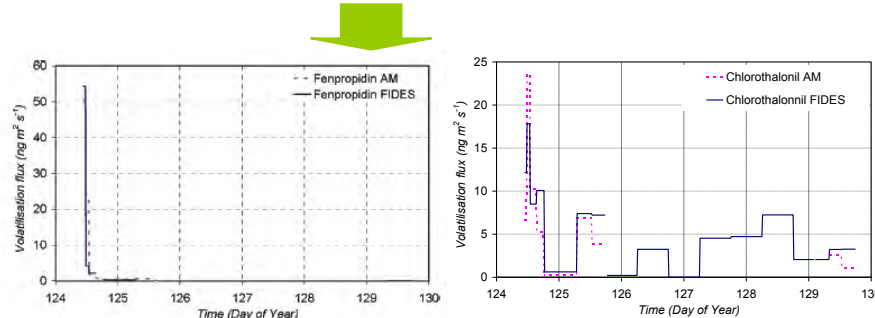
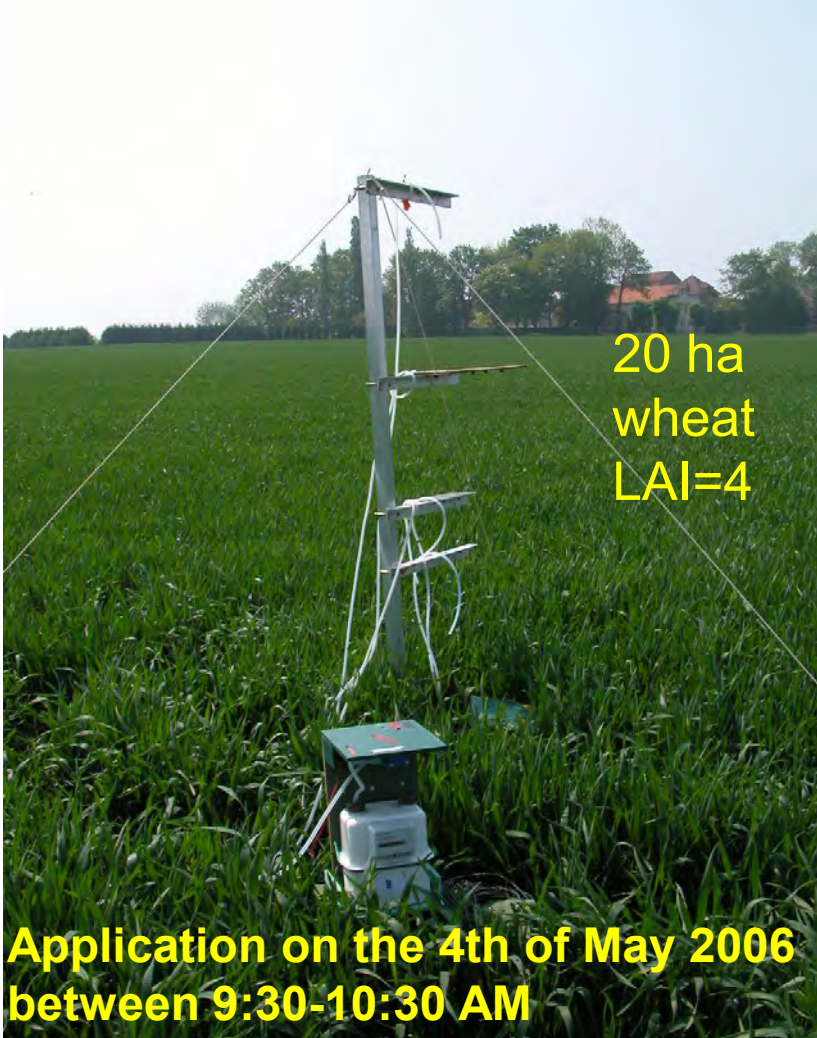
- an energy budget model for soil and leaf surfaces
 - water transfer in the soil considered as a single reservoir with a dry layer at the surface
 - a pollutant exchange model (fluxes of NH_3 , O_3), which distinguishes the soil and leaf exchange processes and which is directly coupled to the energy balance via the soil and leaf surface temperatures
- + interception of the spraying solution by the crop (from Gyldenkaerne et al., 1999) implemented for the purpose of this study

Material and Methods: Experimental set-up (Bedos et al., 2010)

Two fungicides: **Chlorothalonil** ($7.6 \cdot 10^{-5}$ Pa) and **Fenpropidin** ($1.7 \cdot 10^{-2}$ Pa)

Flux measured from May 4 to May 9

**Aerodynamic
+ Inversion method**
Loubet et al. (2010)

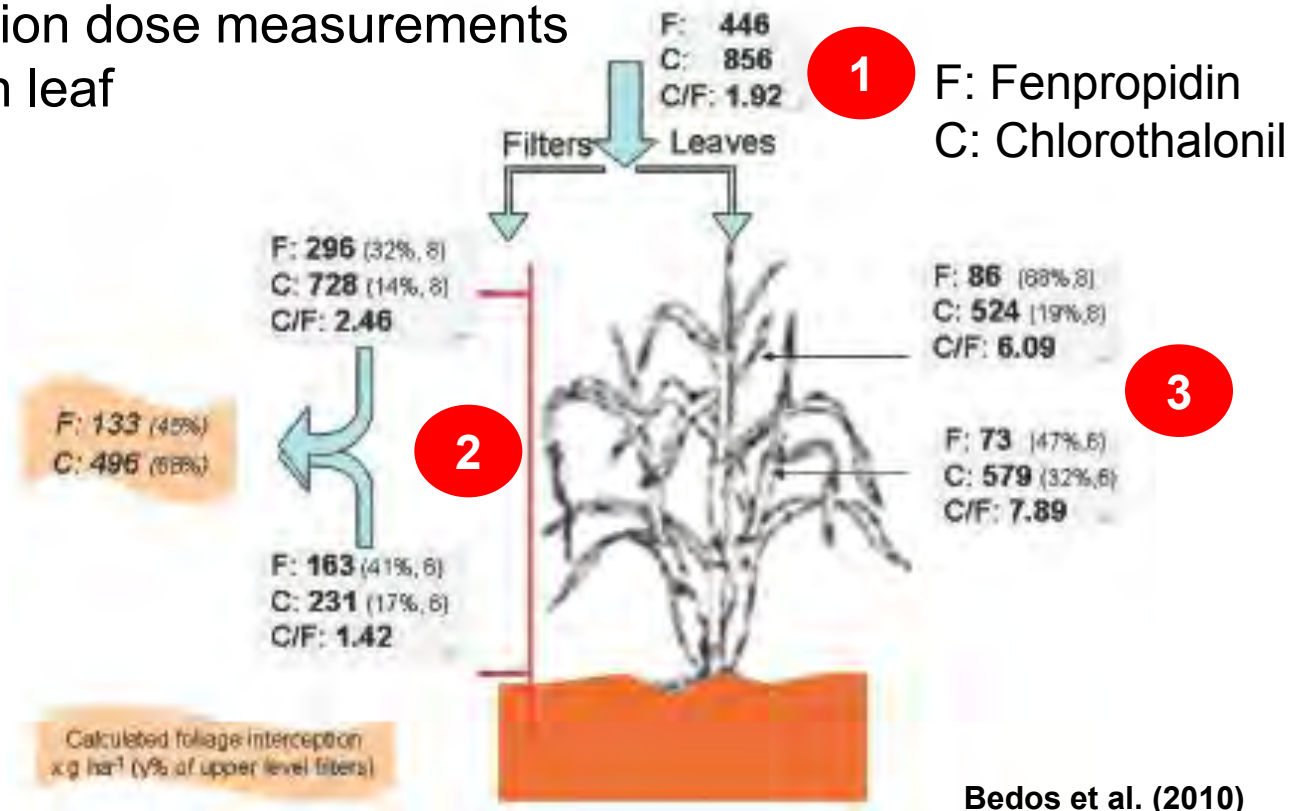


+ micrometeorological conditions:
evaporation, sensible heat flux, leaf
and soil surface temperatures

Material and Methods : Experimental set-up

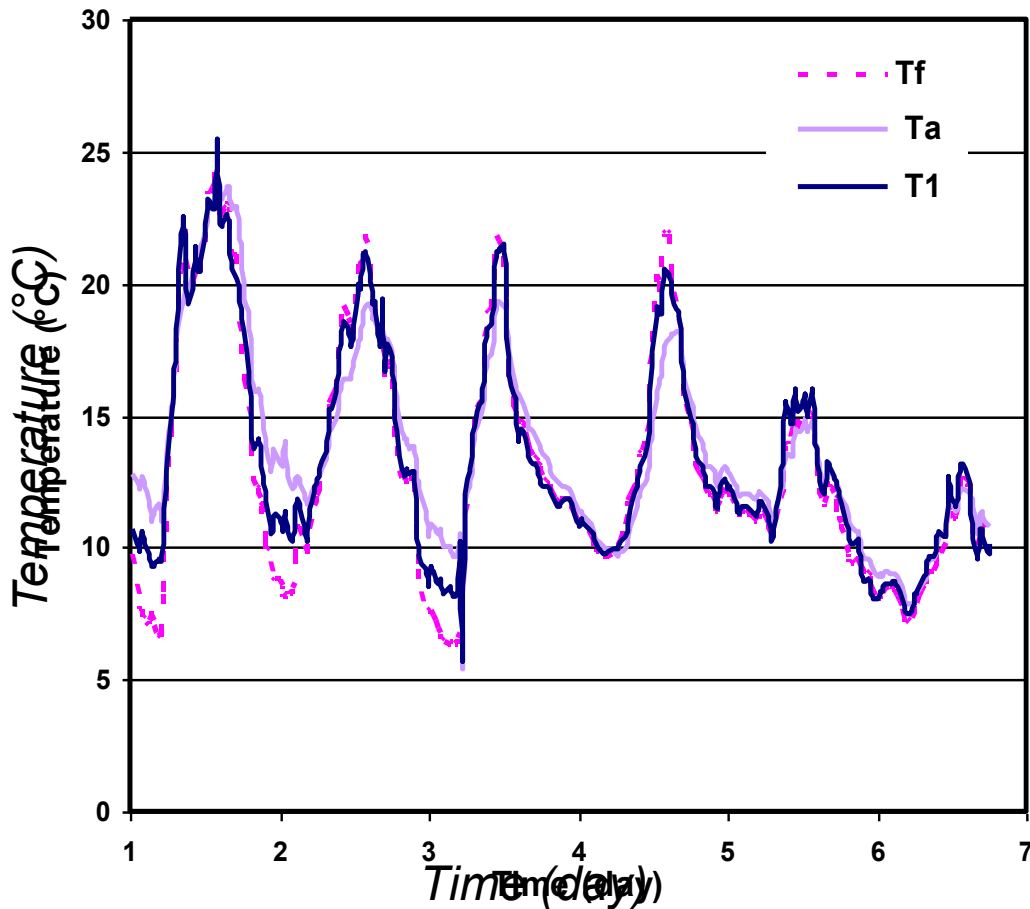
Focus on the application dose measurements
+ pesticide residue on leaf

Initial distribution crop/soil



- ⇒ Interception of the application by the crop : input data for the volatilization model
The model is run with measured applied amount on leaves, with an application assumed at 10:30 (end of the real application)
- ⇒ Need to improve the estimation of the application dose
Cf. Workshop 2008 (Cambridge)

Results Comparison of modelled and measured surface temperature of leaves

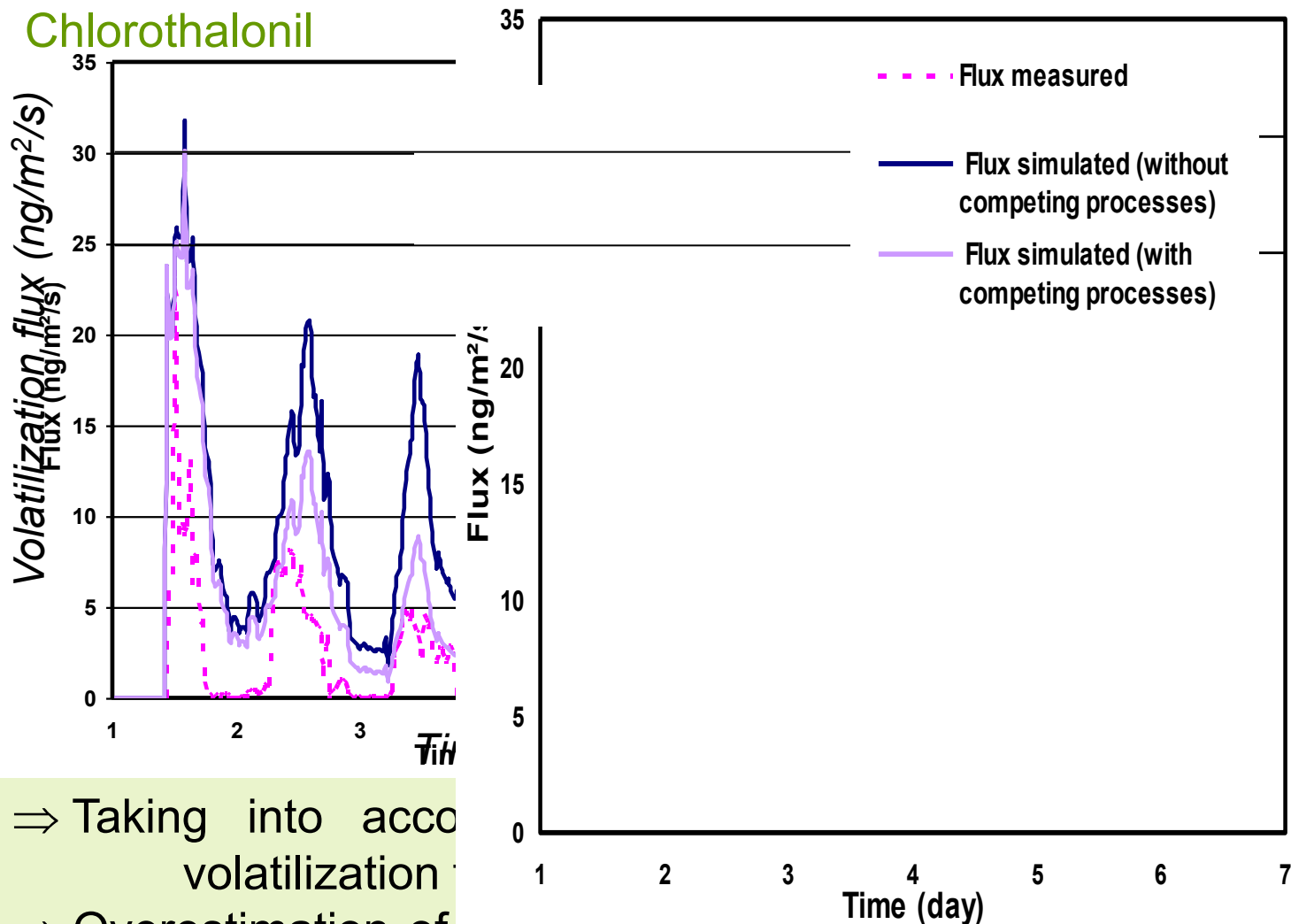


Tf : the measured leaf temperature
Ta : the measured air temperature
T1 : the simulated leaf temperature

⇒ Pretty good agreement

⇒ Leaf temperature and air temperature different ($T_f - T_a = 2^\circ\text{C}$ during daytime)

Results Comparison of modelled and measured flux volatilization of Chlorothalonil



ing processes)

processes)

esses:

lorothalonil
g et al. (2008)

= 0,14 d⁻¹

(ation)= 0,23 d⁻¹

describing the

⇒ Taking into account
volatilization

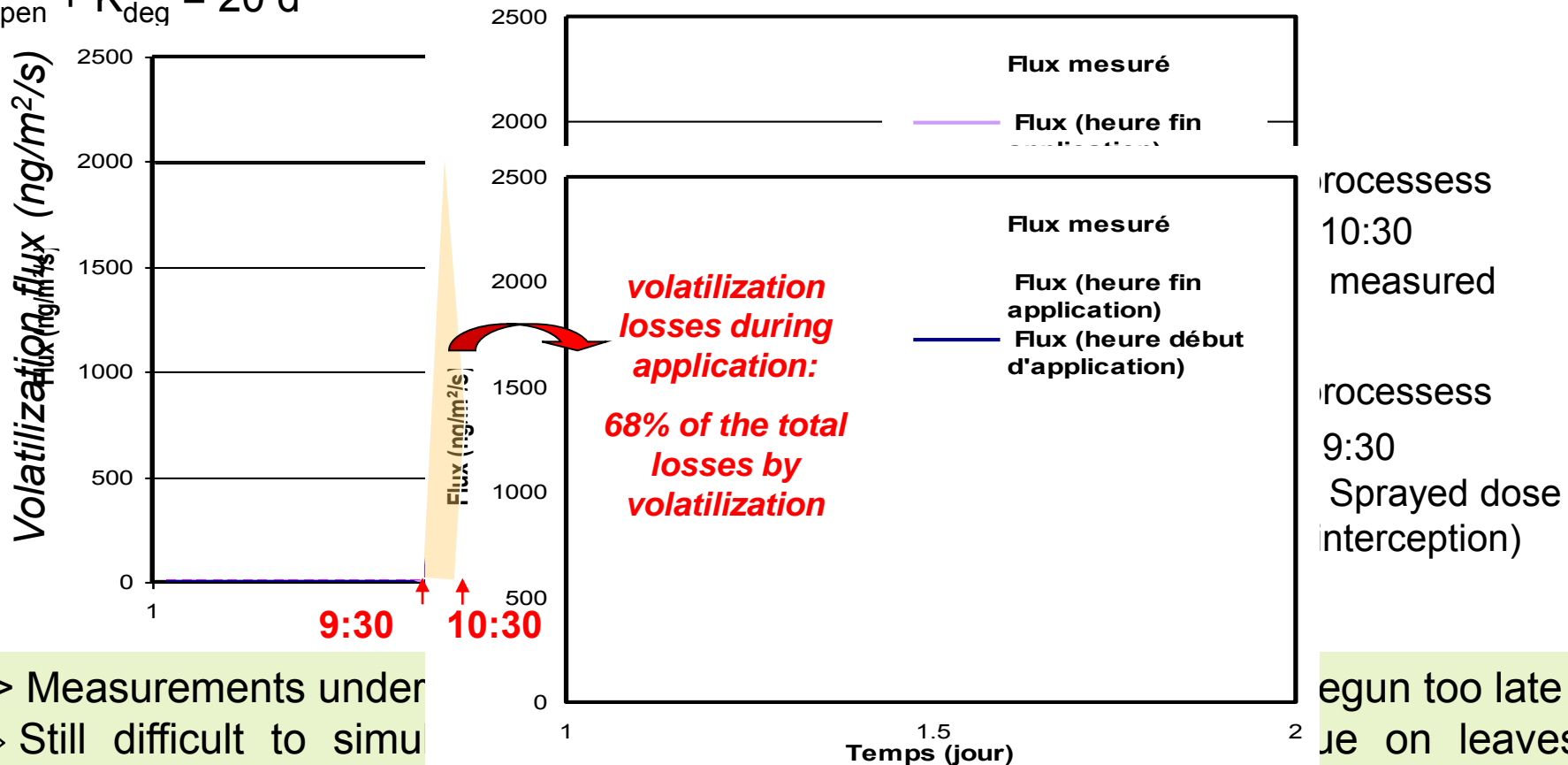
⇒ Overestimation of
ones

early volatilization rates and under-estimation of later

Results: Comparison of modelled and measured flux volatilization of Fenpropidin

No values found in the literature for competing processes, best results found for

$$K_{pen} + K_{deg} = 20 \text{ d}^{-1}$$



=> Measurements under
=> Still difficult to simulate

overestimated at the time of the end of application vs the measured one (not shown) => fast dissipation and/or effect of formulation involved?

processes
10:30
measured
processes
9:30
Sprayed dose
(interception)

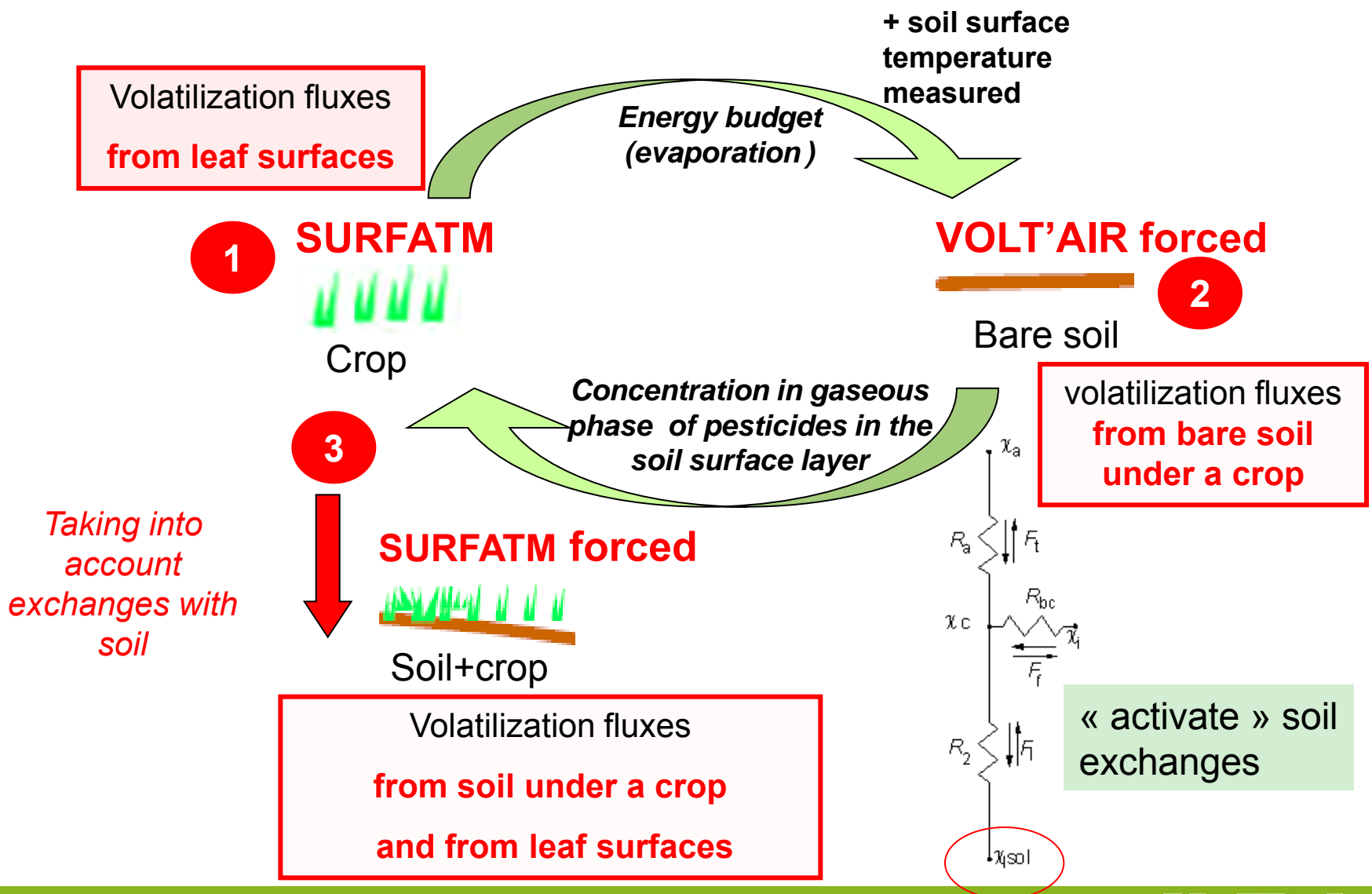
begun too late
due on leaves



Conclusions: volatilization from plant surfaces

- description of the volatilization fluxes is possible when the coefficients for competing processes are known
- to go further on, we have to:
 - Analyse the time evolution of pesticide residue on leaves
 - Mechanistically describe competing processes
 - Measurements : better estimate residue on leaves and early stage volatilization
- Study the contribution of volatilization from soil and from leaves

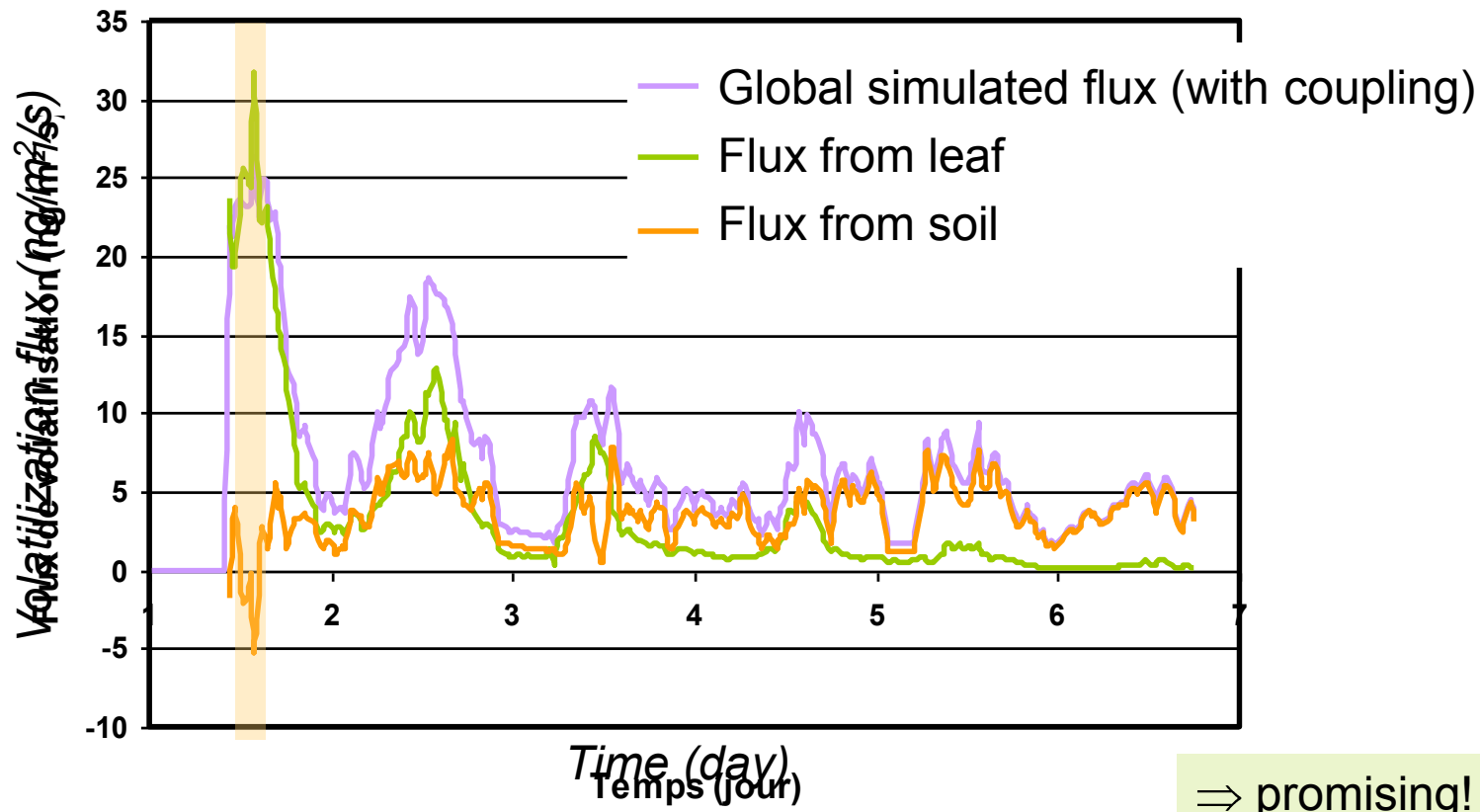
Perspectives: Coupling « off-line » SURFATM and Vol'Air-Pesticides



Perspectives: Coupling « off-line » SURFATM and Vol'Air-Pesticides

First results:

- * contribution of volatilization from soil and from leaf surfaces as a function of time
- * Deposition on soil just after application



⇒ promising!



Thank you for your attention

Special thanks to E. Van Den Berg for his help on the parameterization of volatilization