

The role of single particle ICP-MS with microsecond time resolution in a multi-technique approach for unveiling the transformations of ingested metal-based nanoparticles

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Peculiar fate of ingested nanomaterials

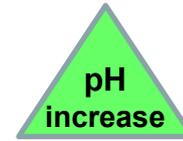
Nanoparticles entering the human body via the oral route are subjected to **conditions that are very different from those encountered via other exposure routes**

Physicochemical properties of any ingested particulate material will be affected by:

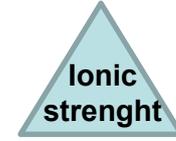
- extreme **pH** and **ionic strength** shifts encountered during gut transit



Stomach



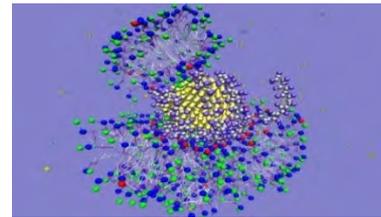
Small intestine



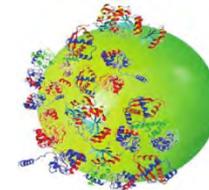
Stomach & intestine

- the **co-ingested food material**

Interaction between particles and food components



- the **enzymes and bile acids secreted** within the gut



Changes promoted by interaction with the GI environment

The extremely **low pH** of the stomach may promote:



oxidation/dissolution of nanoparticles constituted by soluble metals/metal oxides with **release of constituent ions**

The **high ionic strength** in the stomach and intestine critically affect nanomaterial properties, in many cases leading to:



agglomeration in the gastric phase



deagglomeration in the intestinal phase

Important factors affecting **degradation/dissolution** and **agglomeration/deagglomeration** are expected to include **physical forces, temperature, pH, presence of enzymes, salts and bile, interactions with food components**

In vitro digestion models to assess the fate of ingested NMs

Changes in:
size-related properties
shape
surface characteristics

May affect the:
intestinal uptake of
the particles
kinetic behaviour of
the particles

Overall, interactions of nanoparticles with the gastrointestinal environment critically affect their **biological and toxicological properties**

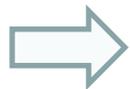


***Acellular in vitro* methods simulating human digestion and mimicking physiological conditions *in vivo* as tools to assess the modifications/dissolution of nanomaterials**



There is a **lack of validation** and **standardisation** of *in vitro* digestion models for nanomaterials

Time-dependent transformations of real world particulate materials under conditions representative of the gastrointestinal tract have been studied by robust *in vitro* digestion methods **with standardised protocols** in the:



Nanomaterials that **quickly dissolve/degrade** in the gastrointestinal tract do not give rise to nano-specific concerns and **standard risk assessment can be followed**



If nanoparticles persist as such after gastrointestinal digestion **they may be absorbed in the gut** and **nanospecific risk assessment is required**



In vitro simulated GI digestion

Assessment of the degradation rate of nanomaterials in conditions representative of the human gastrointestinal tract is considered the **key first step**

An *in vitro* digestion method suitable for food under **fed conditions** has been described by Minekus *et al.* (2014)



FED

Model food + nanoparticles



FASTED

Dispersion of nanoparticles

Less representative of the use of nanomaterials in food products (but may be better for food supplements): low pH conditions in the stomach may promote the dissolution of most metals and metal oxides

Oral phase

Salivary amylase, 2 min, pH 7

Gastric phase

Gastric fluid + pepsin, 2 h, pH 3

Intestinal phase

Intestinal fluid and enzymes, 2h, pH 7

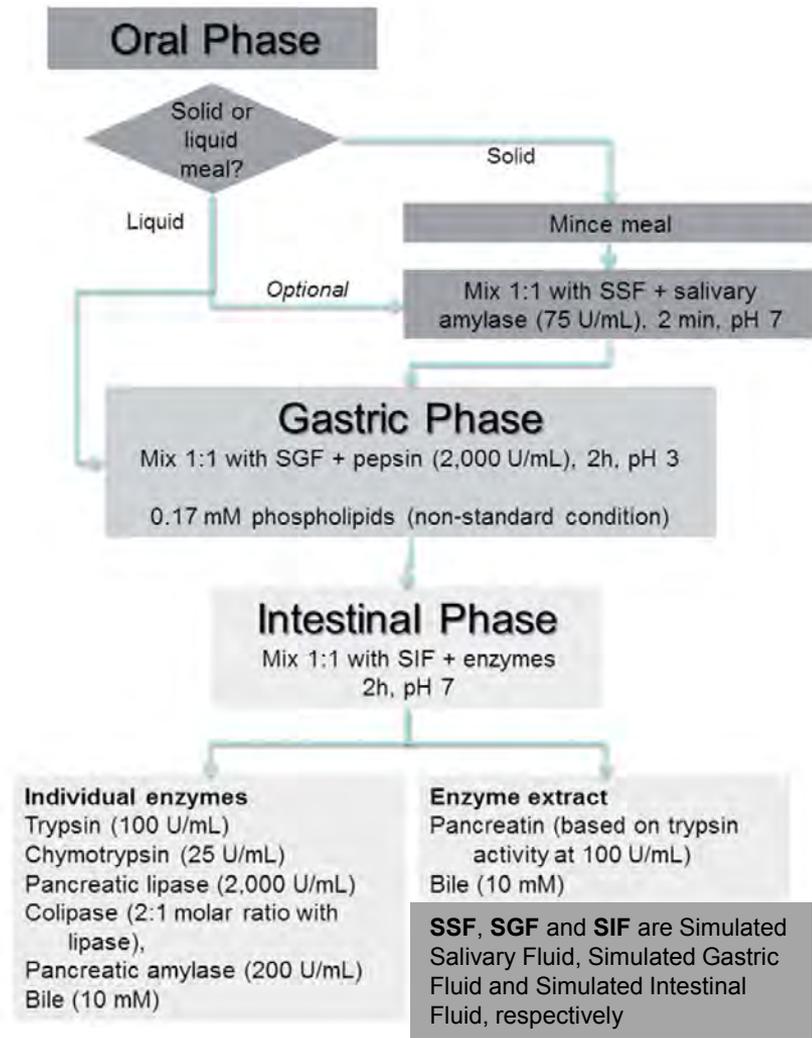


Representative of the physiological state

Sampling

Pristine material + Material in the food matrix (before digestion)

Material after intestinal phase
 Time points: **5, 15, 30 and 60 min**



Mouth

Stomach

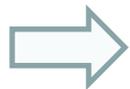
Small intestine



Minekus et al. (2014)
 Food Funct. 5:1113



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EFSA GUIDANCE – *in vitro* simulated GI digestion



A nanomaterial is considered to **degrade quickly** (i.e. have a high degradation rate) if the degradation rate profile in the intestinal phase shows a **clear decrease in the presence of particles over time (no plateau)**, and that **12% or less of the material (mass-based)** – compared with the particulate concentration at the beginning of the *in vitro* digestion – is present as particles **after 30 min of intestinal digestion**



As a sub-argument, it is also assumed that even if a fraction of such quickly degrading materials is absorbed as particles, it **is expected that further degradation will occur under e.g. lysosomal conditions** and that they are unlikely to remain as particles for a long time

The nanomaterials studied:

Ten different food-grade particulate materials belonging to four chemical classes were studied:

- **Synthetic amorphous silica** 4 samples of the food additive (E551): 2 representative of food-related applications from the JRC repository, 2 from commercial suppliers
- **Titanium dioxide** 2 different samples of the food color (E171) from commercial suppliers
- **Iron oxides/hydroxides** 2 samples of the food color (E172) from commercial suppliers: one Yellow iron oxide (FeO(OH)) and one Red iron oxide (Fe₂O₃)
- **Zinc oxide** 2 samples: one used as nutrient source for human consumption, the other one for animal consumption

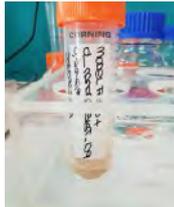
The experimental procedure

Cereal based model food

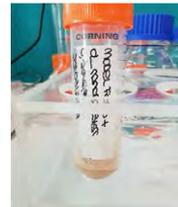
Composition of model food:

- Fats: 5%
- Carbohydrates : 11%
- Fiber: 3%
- Proteins: 11%
- Salt: 2%

The model food was homogenized and a part was stored for further studies



2'
Salivary SF



2h
Gastric SF

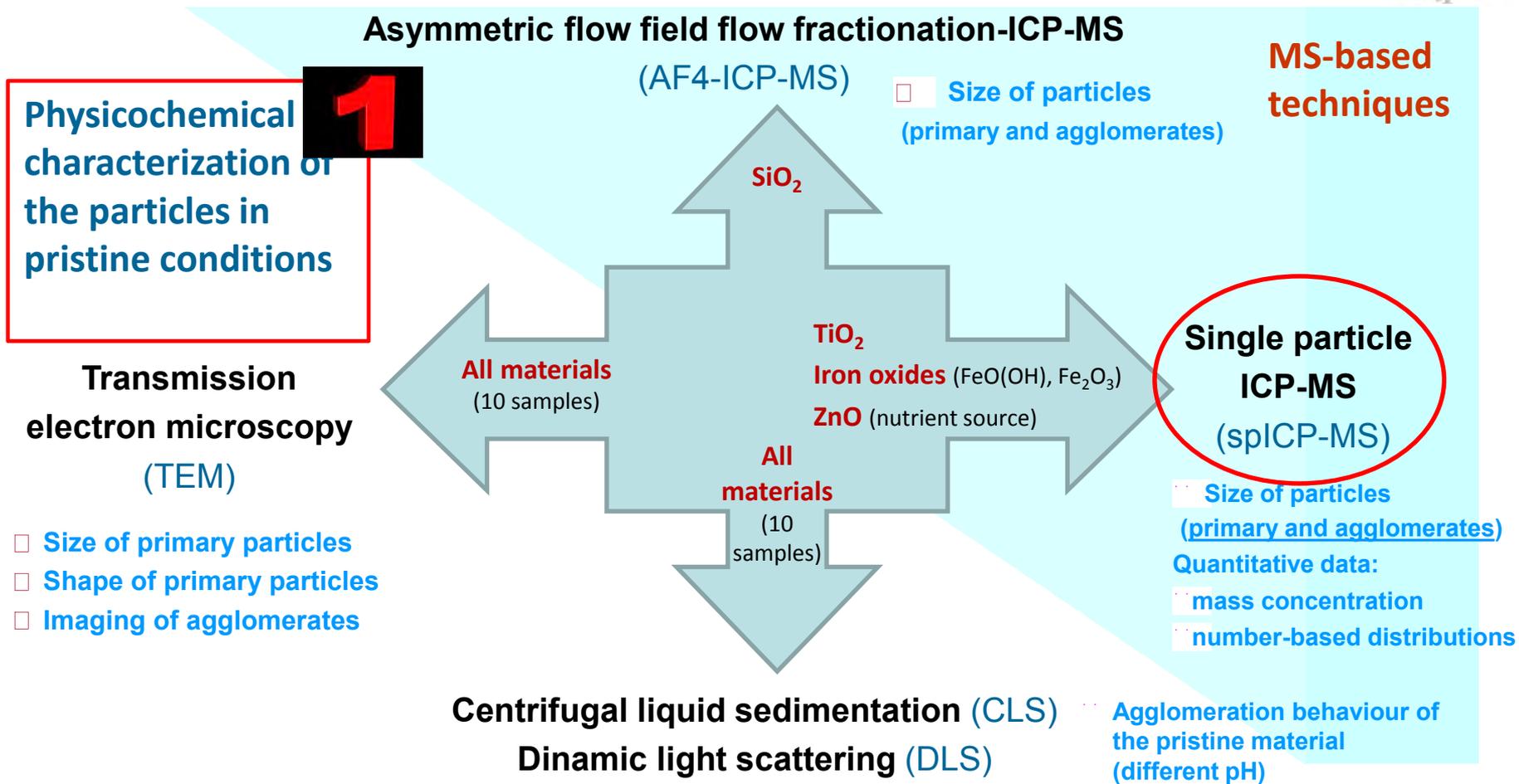


30'
Intestinal SF



Prior in vitro digestion, the model food was spiked with a nanoparticles dispersion and **incubated for at least one hour at room temperature**

Physicochemical characterization: multi-method approach



Titanium dioxide (TiO₂)

Similar results obtained by spICP-MS : **small agglomerates** in addition to primaries (visible by **higher frequencies at larger diameters**)

Physicochemical characterization: multi-method approach

Characterization

2

Asymmetric flow field flow fractionation-ICP-MS

After ingestion: Fed and fasted conditions

After lysosomal processing

(AF4-ICP-MS)

□ Size of particles (primary and agglomerates)

MS-based techniques

SiO₂

TiO₂

Iron oxides (FeO(OH), Fe₂O₃)

ZnO (nutrient source)

Single particle ICP-MS

(spICP-MS)

Transmission electron microscopy (TEM)

All materials (10 samples)

□ Size of particles (primary and agglomerates)

Quantitative data:

□ mass concentration

□ number-based distributions

- Size of primary particles
- Shape of primary particles
- Imaging of agglomerates

All materials (10 samples)

Centrifugal liquid sedimentation (CLS)

Dinamic light scattering (DLS)

Agglomeration behaviour of the pristine material (different pH)

Preliminary results post gastrointestinal *in vitro* digestion- sp ICP-MS

Similar to the distribution obtained for the characterization of **pristine TiO₂** by sp ICP-MS

Conclusions

- **Nanoparticles** entering the human body via the **oral route** are subjected to conditions that are very different from those encountered via other exposure routes
- Physicochemical properties of ingested particulate material are affected by **pH** and **ionic strength shifts, co-ingested food material, enzymes** secreted within the gut. This environment critically affect the **biological** and **toxicological** properties of the ingested nanoparticles
- Time-dependent transformations of **real world particulate** materials applying the in vitro acellular methods were studied according to the assessment scheme of the EFSA Guidance
- **Ten** different **food-grade particulate materials** belonging to **four chemical classes** were studied and the focus was on checking **robustness of protocols** for possible standardisation
- A state-of-the-art **multi-technique** approach was used for the **physicochemical characterization** of the materials before and after the tests
- **Single particle ICP-MS** as an **essential technique** in order to characterize nanomaterials after the simulated *in vitro* GI digestion

Acknowledgements

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THANK YOU FOR YOUR ATTENTION !

